

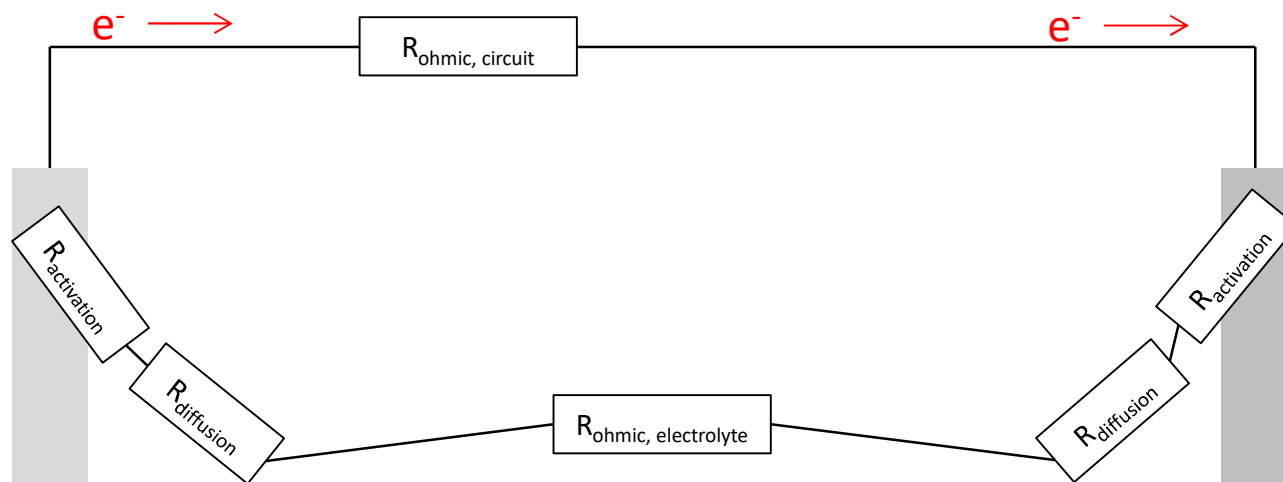
Microbial electrochemical technologies – fundamental aspects and methods of analysis

Week 7 – Methods and analysis 3

- Electrochemical impedance spectroscopy
 - The method
 - Equivalent circuit modelling
 - Applications in studies of METs

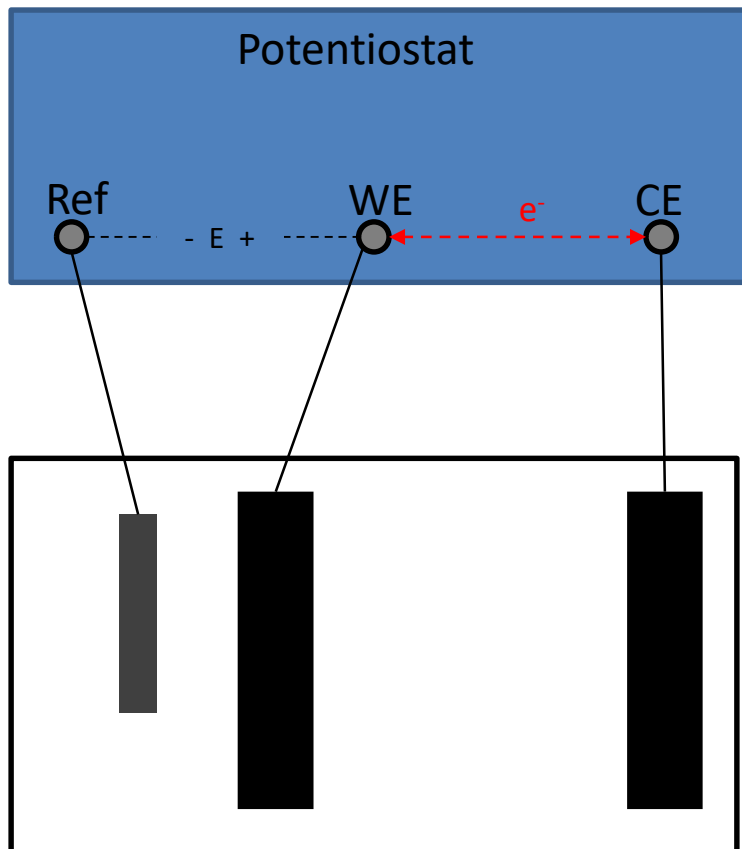
Before we start 1...

- Remember, the figure below is a simplified model. Capacitors at the electrode surfaces could be added to explain non-faradaic currents during potential steps. Observed resistances also depend on e.g. mass transfer limitations.

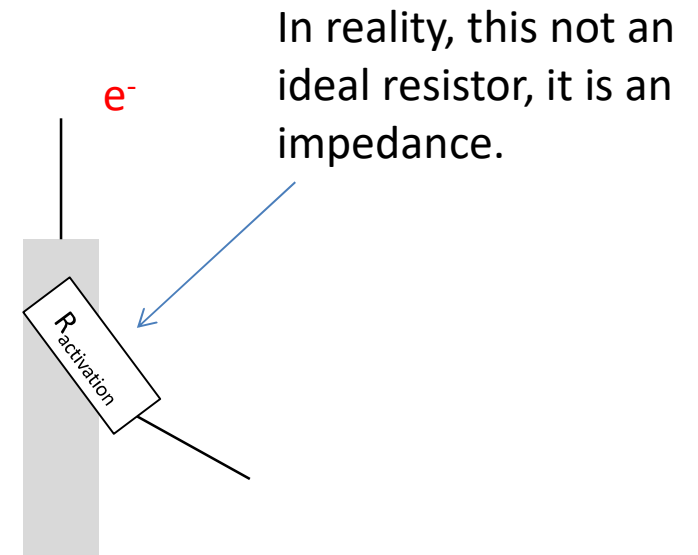
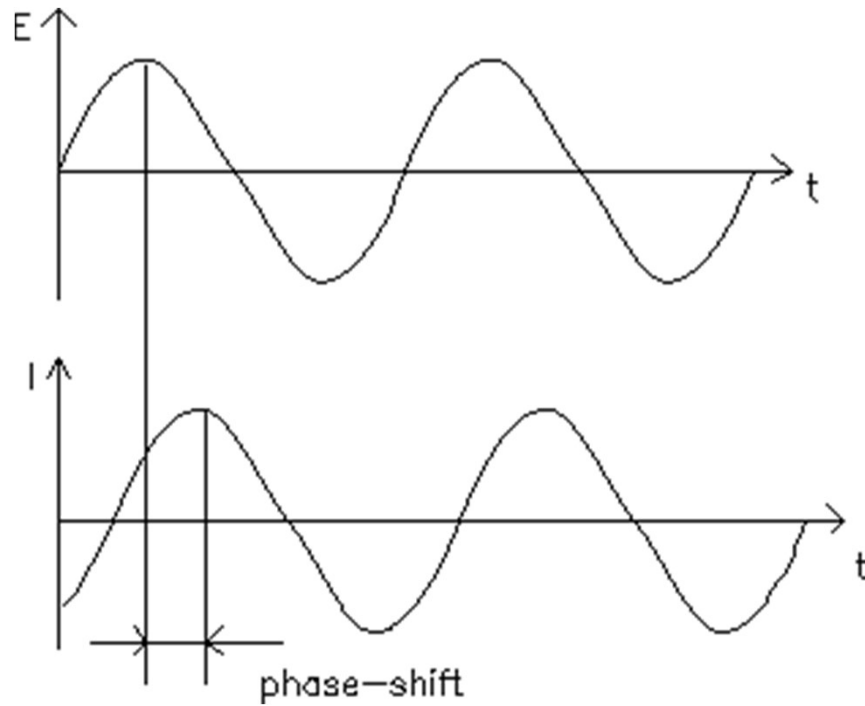


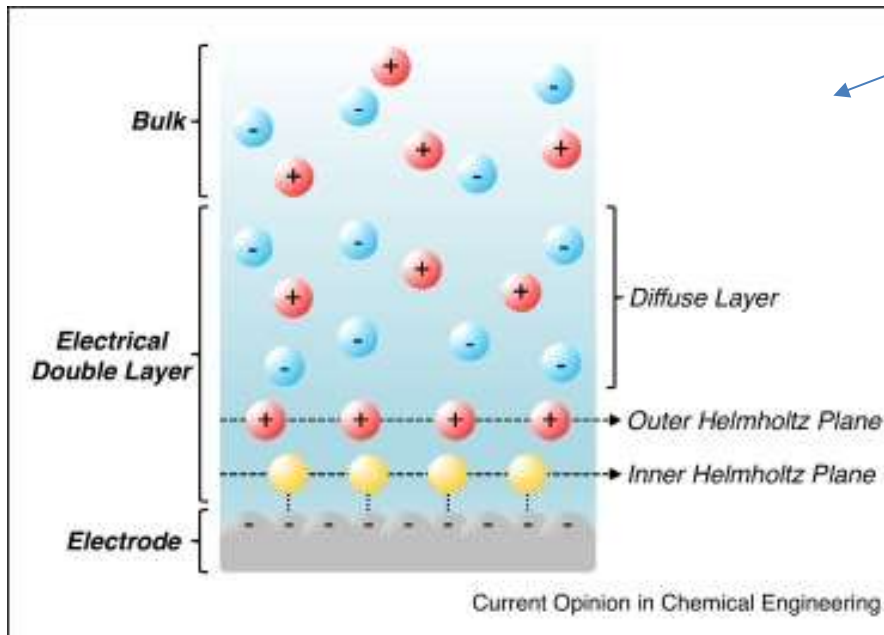
Before we start 2...

- Remember how the potentiostat functions

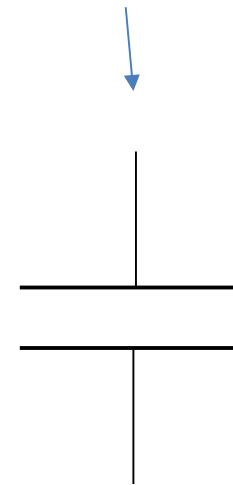


The method



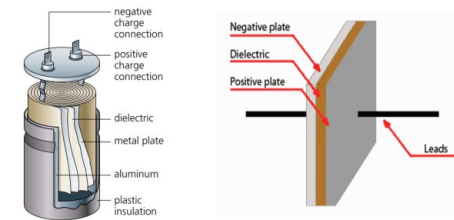


Similar to a capacitor



Dunwell et al. (2018). Current opinion in Chemical Engineering, 20, 151-158.

Capacitor Construction



Impedance

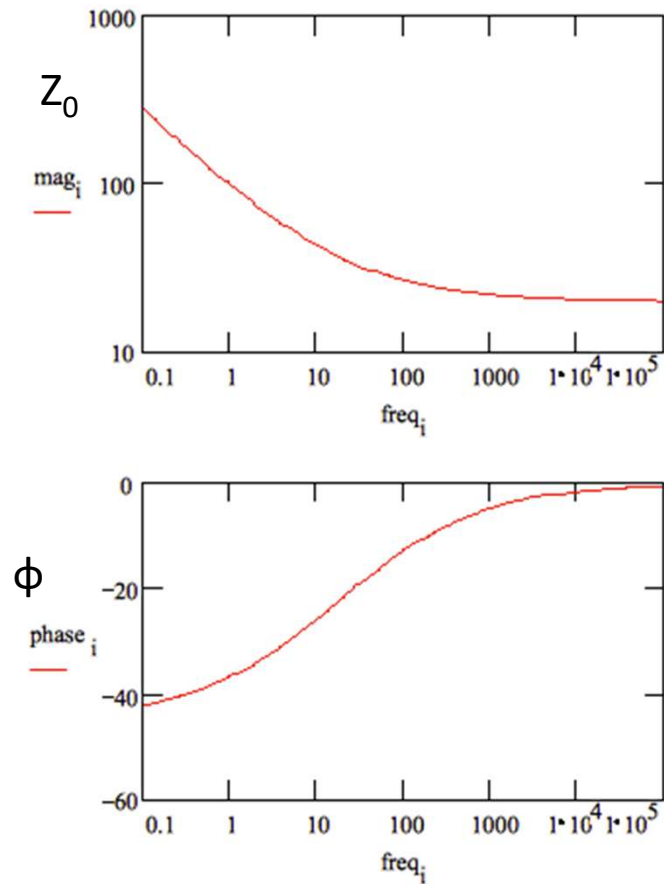
- $E_t = E_0 \sin(\omega t)$
- $I_t = I_0 \sin(\omega t + \phi)$
- $Z = \frac{E_t}{I_t} = \frac{E_0 \sin(\omega t)}{I_0 \sin(\omega t + \phi)} = Z_0 \frac{\sin(\omega t)}{\sin(\omega t + \phi)}$
- The impedance is a function of a magnitude (Z_0) and a phase shift (ϕ)
- It can also be represented as a complex number:
- $Z(\omega) = Z_0 (\cos(\phi) + j \sin(\phi))$

ω = radial frequency (rad/s), $\omega = 2\pi f$

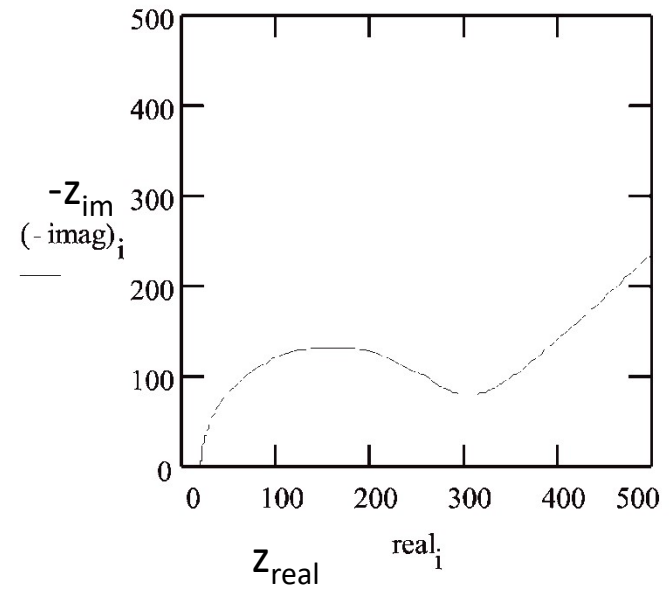
Obtaining data

- Determine Z_0 and ϕ for different frequencies

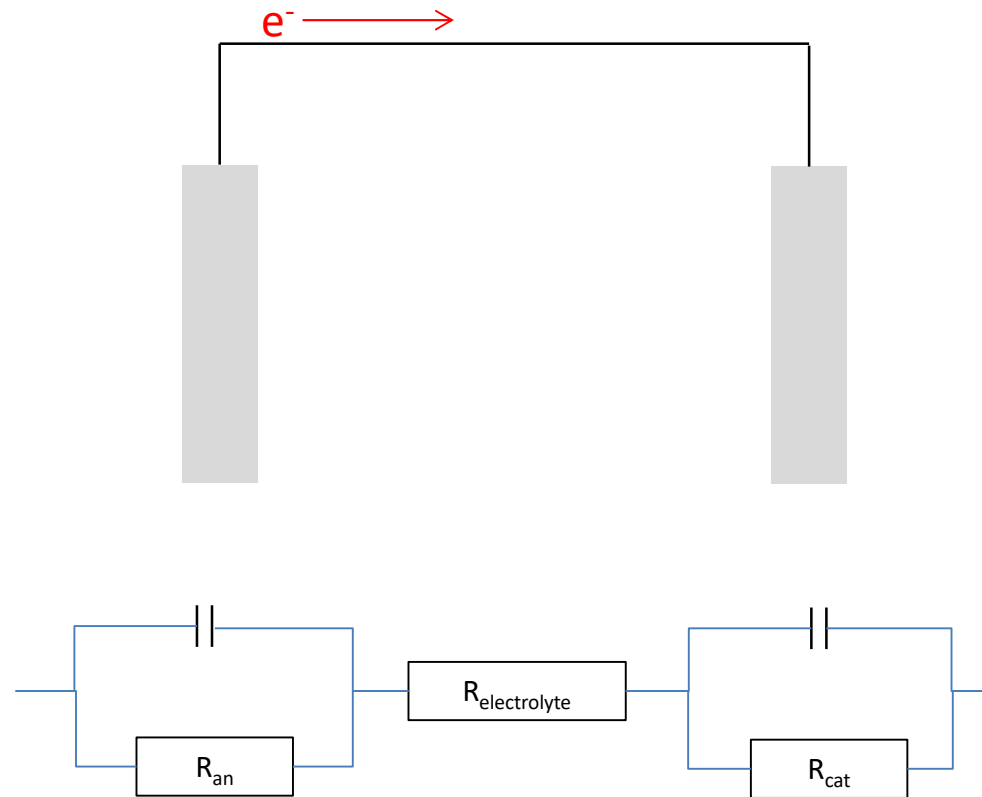
Bode plot



Nyquist plot

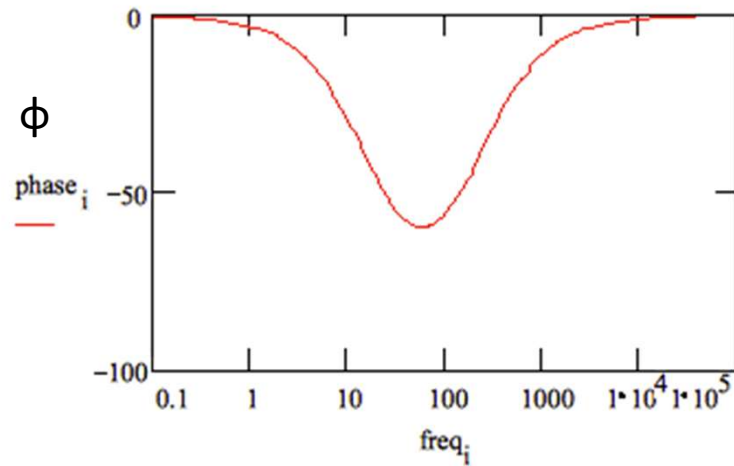
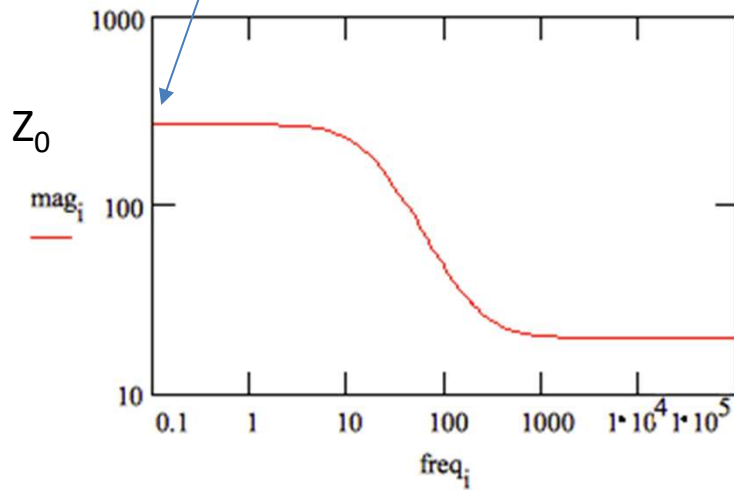


Equivalent circuit modelling



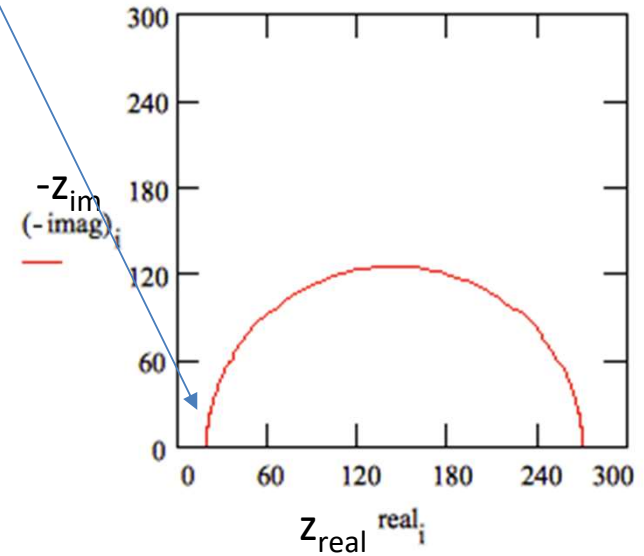
<http://www.gamry.com/application-notes/basics-of-electrochemical-impedance-spectroscopy/>

Low freq = capacitor impedance infinite. Thus $Z_0 \approx R_{an} + R_{cat} + R_{electrolyte}$

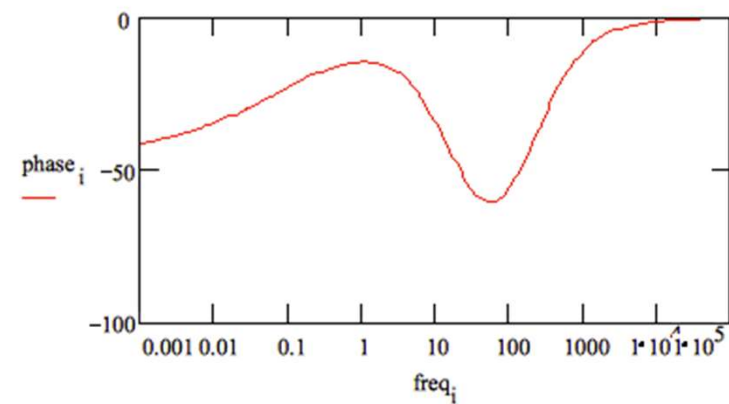
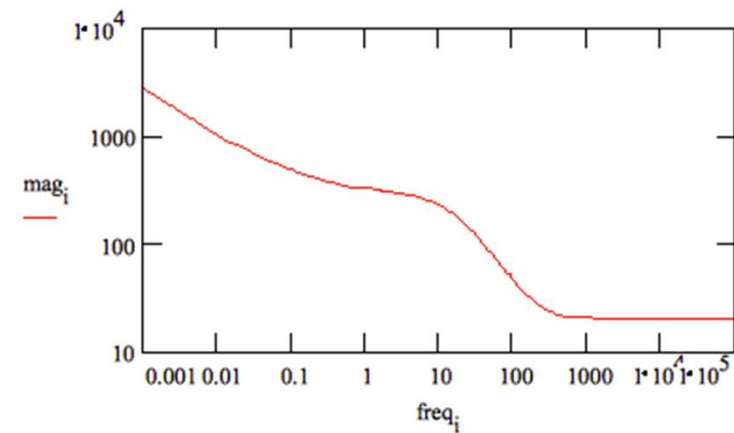
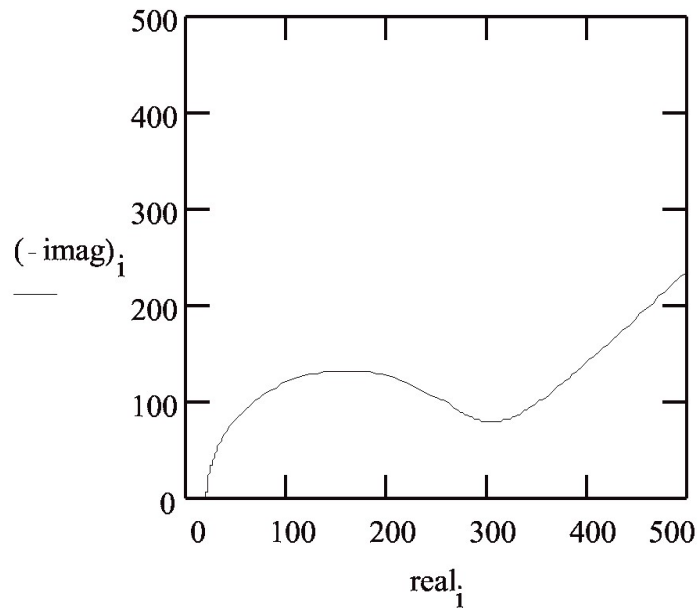
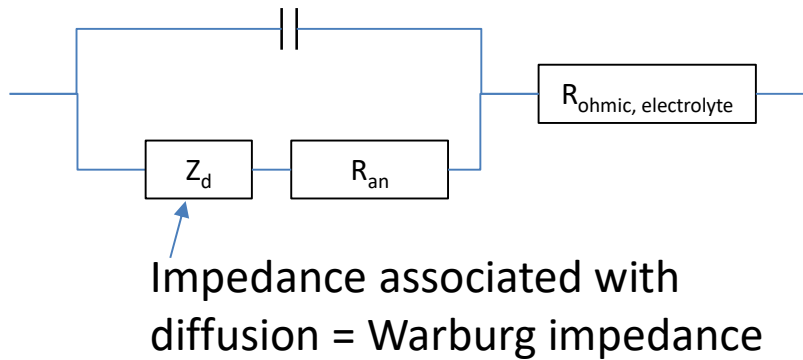


High freq = almost no impedance associated with capacitor. Thus

$$Z_0 \approx R_{electrolyte}$$



Circuit model including diffusion



Some examples

Aaron et al. *Energies*, 3, 592-606, 2010

Ter Heijne et al. 2011, *Energy & Environmental Science* 11(4), 5035-5043.